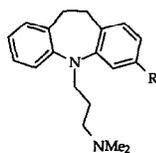


## A Free Radical Route to the Benzazepines and Dibenzazepines

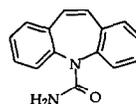
Zhizhen Barbara Zheng and Paul Dowd\*  
Department of Chemistry, University of Pittsburgh  
Pittsburgh, PA 15260

**Abstract** Free radical ring expansion of six membered azocycles provides a new entry to the preparation of benzazepines and dibenzazepines.

Benzazepines and dibenzazepines have important pharmacological activity. Dibenzazepines include imipramine and clomipramine, substances widely used for the treatment of depressive illness.<sup>1,2,3</sup> The 5H-dibenz[b,f]azepine-5-carboxamide (carbamazepine) is a well known



Imipramine, R=H  
Clomipramine, R=Cl

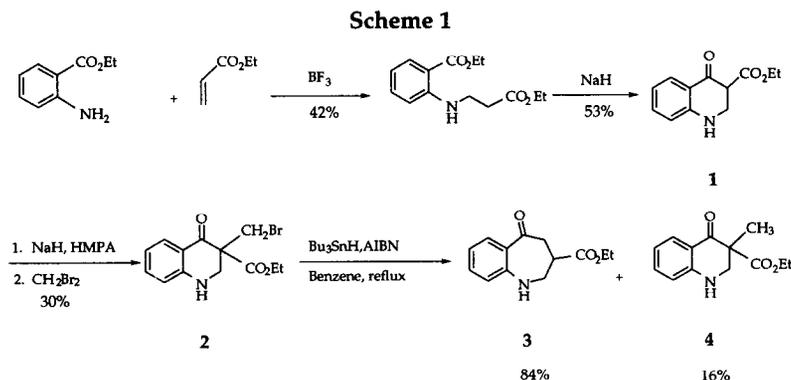


Carbamazepine

analgesic and anticonvulsant. In this paper, we report the free radical ring expansion of six-membered azocycles leading to the synthesis of benzazepines and dibenzazepines by a novel route. The intermediate seven-membered benzazepine  $\gamma$ -keto ester in this synthesis may also be of value in the synthesis of alkaloids and other products with biologically useful properties.

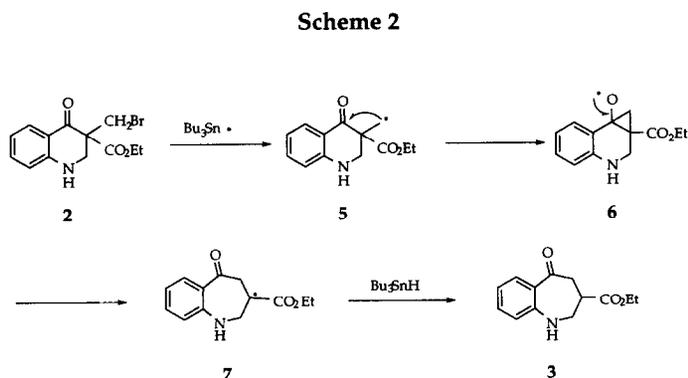
For the benzazepines, preparation of the required  $\beta$ -keto ester **1** makes use of a Dieckmann condensation of the Michael adduct of ethyl anthranilate with ethyl acrylate<sup>4</sup> (Scheme 1). Alkylation of **1** with dibromomethane and sodium hydride in refluxing

tetrahydrofuran yielded the adduct **2** (Scheme 1). Tri-*n*-butyltin hydride was added to a refluxing



benzene solution of **2** using a syringe pump over a 10 h period with a catalytic amount of AIBN. The ring expansion product **3**<sup>5</sup> was obtained (84%) with a minor amount of the direct reduction product **4** (16%).

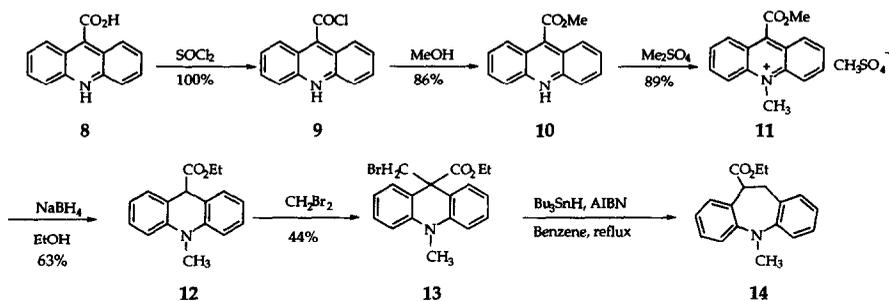
Following earlier examples of free radical ring expansion<sup>6</sup>, it is reasonable to suggest that formation of the primary radical **5** is followed by attack on the ketone (Scheme 2). The resulting



alkoxyl radical then undergoes regiospecific ring-opening leading to the ester-stabilized radical **7**.

For the dibenzazepines, the starting 9-acridinecarboxylic acid **8** was converted to its ester according to a literature procedure<sup>3,7</sup> (Scheme 3). Thus, treatment of **8** with thionyl chloride yielded the acid chloride **9** (100%) which was then converted to the corresponding methyl ester **10** by methanolysis (86%). Methylation of **10** with dimethyl sulfate yielded the quaternary amine salt **11** (89%), which was then reduced to **12** with sodium borohydride (63%).

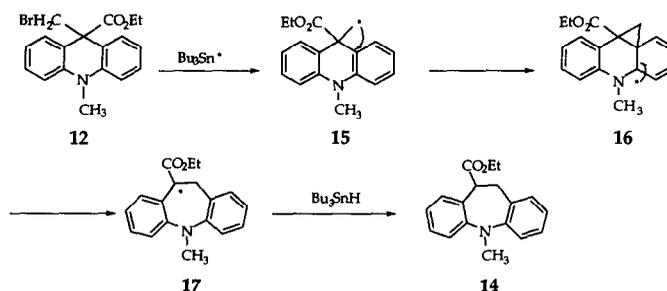
Scheme 3



Alkylation of **12** with dibromomethane and sodium hydride in tetrahydrofuran at room temperature for 16 h yielded **13**, which was treated in refluxing benzene with AIBN and  $\text{Bu}_3\text{SnH}$  (syringe pump, 10 h) and led smoothly to the ring expansion product **14** (68%).<sup>8</sup>

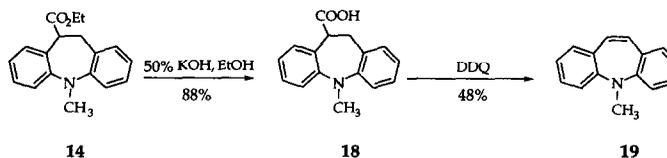
Rearrangement of the primary radical **15** is assisted by the formation of the conjugated radical **16**.<sup>9</sup> Opening of **16** leads to the more stable radical **17**, which can then abstract a hydrogen atom from  $\text{Bu}_3\text{SnH}$  to produce the desired product **14** (Scheme 4).

Scheme 4

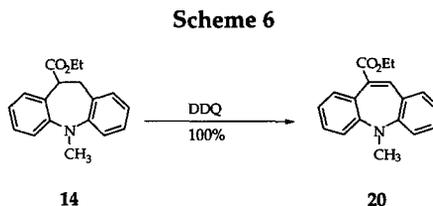


The ester group of **14** can be removed, eliminated or converted to other functional groups and a further series of dibenzazepines can be prepared. For example, after hydrolysis of ethyl ester **14**, the resulting acid **18** was decarboxylated to the corresponding *N*-methyl dibenzazepines **19** (Scheme 5) by treatment with DDQ.

Scheme 5



DDQ was also used to dehydrogenate **14** to **20** (Scheme 6), maintaining the skeleton of the



dibenzazepine with the additional ester available for transformation to other functional groups.

Since the substituent on nitrogen can readily be varied, the free radical route provides a general, clean, high-yield route to the synthesis of benzazepines and dibenzazepines.

### Acknowledgement

This work was generously supported by the Institute of General Medical Sciences of the National Institutes of Health under grant GM 39825.

### References

1. For a review and pertinent references see: R. K. Smalley in *"Comprehensive Heterocyclic Chemistry"* Vol. 7, W. Lwowski, Ed., Pergamon Press, Oxford, 1984, pp. 491-546.
2. Kricka, L. J.; Ledwith, A.; *Chem. Rev.* **1974**, *74*, 101.
3. Allais, A.; Guillaume, J.; Poittevin, A.; Nedelec, L.; Chiffot, L.; Peterfalvi, M.; Hunt, P. *Eur. J. Med. Chem.* **1982**, *17*, 371.
4. Proctor, G. R.; Ross, W.L.; Tapia, A. *J. Chem. Soc., Perkin Trans I* **1972**, 1803.
5. Compound **3**:  $^1\text{H-NMR-CDCl}_3$  (ppm): 7.70 (d, 1H), 7.25 (t, 1H), 6.85 (t, 1H), 6.70 (d, 1H), 4.65 (s, 1H), 4.2 (q, 2H), 3.85 (m, 1H), 3.25 (m, 3H), 2.95 (m, 1H), 1.25 (t, 3H).  $^{13}\text{C-NMR-CDCl}_3$  (ppm): 200.2, 193.0, 153.3, 132.8, 129.6, 124.9, 119.1, 117.6, 61.3, 50.1, 47.8, 43.0, 14.2. IR (KBr,  $\text{cm}^{-1}$ ): 3352.7 (vs, N-H), 1728.4 (vs, C=O), 1645.5 (vs, C=O). MS (m/e): 233 ( $\text{M}^+$ , 50), 188 (20), 160 (30), 133 (100), 105 (40). mp 64-65.0  $^\circ\text{C}$ .
6. For recent review, see: Dowd, P.; Zhang, W., "Free Radical Mediated Ring Expansion and Related Annulations." *Chem. Rev.*, in press. For free radical ring expansion of  $\beta$ -keto esters, see: Dowd, P.; Choi, S.-C., *J. Am. Chem. Soc.* **1987**, *109*, 3493. Dowd, P.; Choi, S.-C., *J. Am. Chem. Soc.* **1987**, *109*, 6548. Dowd, P.; Choi, S.-C., *Tetrahedron Lett.* **1989**, *45*, 77. Dowd, P.; Choi, S.-C., *Tetrahedron* **1991**, *47*, 4847. Beckwith, A. J.; O'Shea, D. M.; Gerba, S.; Westwood, S. W.; *J. Chem. Soc., Chem. Commun.* **1987**, 666. Beckwith, A. J.; O'Shea, D. M.; Westwood, S. W.; *J. Am. Chem. Soc.* **1988**, *110*, 2565. Bowman, W. R.; Westlake, P. J. *Tetrahedron* **1992**, *48*, 4027.
7. Rauhut, M. M.; Sheehan, D.; Clarke, R. A.; Roberts, B. G.; A.M. Semsel, B. G. *J. Org. Chem.* **1965**, *30*, 3587.
8. Compound **10**:  $^1\text{H-NMR-CDCl}_3$  (ppm) 7.39-6.31 (m, ArH, 8H), 4.14 (m, 3H), 3.43(q, 1H), 3.36 (m, 1H), 3.34 (s, 3H), 1.19 (t, 3H).  $^{13}\text{CNMR-CDCl}_3$  (ppm): 173.6, 148.7, 148.1, 131.7, 130.3, 129.8, 129.2, 127.7, 126.8, 122.6, 122.0, 119.6, 118.3, 60.7, 47.9, 39.7, 34.7, 14.3. IR (KBr,  $\text{cm}^{-1}$ ): 1728.4 (vs, C=O). MS (m/e): 281 (40), 208 (100), 193 (60), 179 (10), 65 (10). m.p. 62-62.5  $^\circ\text{C}$ .
9. Sato, T.; Ishibashi, H.; Ikeda, M. *J. Chem. Soc., Perkin Trans. I* **1991**, 353. Curran, D. P.; Liu, H. *J. Am. Chem. Soc.* **1991**, *113*, 27. McNab, H. *J. Chem. Soc., Chem. Commun.* **1990**, 543. Rigby, J. H.; Qabar, M. N. *J. Org. Chem.* **1993**, *58*, 4473.

(Received in USA 30 August 1993; accepted 23 September 1993)